

The role of reaction time in organic phase on the preparation of thin-film composite nanofiltration (TFC-NF) membrane for dye removal

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ABSTRACT

Dyes are widely used in textile-based industries and inefficiencies in the dyeing process may result in large amounts of dyestuff being discharged into the wastewater. Membrane separation technology provides relatively high removal for dye wastewater. In this study, local thin-film composite nanofiltration (TFC-NF) membranes for dye removal were fabricated. These membranes were synthesised using an interfacial polymerization technique by reacting *m*-phenylene diamine (MPD) (2.0% w/w) in an aqueous solution and trimesoyl chloride (TMC) (0.2% w/w) in an organic solution. The role of reaction time (5 s, 10 s, 15 s, 20 s, 30 s, 45 s and 60 s) in organic phase on the morphologies, structural details and performances were examined. Membranes were characterized in terms of permeability coefficient, separation of charged solute, membrane structural details and membrane morphology. All fabricated membrane characteristics were in the range of nanofiltration (NF) membranes. The membrane that was prepared with the reaction time 30 s in organic phase (TFC-NF-R30) revealed the highest perm-selective performance due to the trade-off between rejection (87%) and flux ($8.65 \times 10^{-6} \text{ m}^3/\text{m}^2\text{s}$) of charged solute. The membrane was further analysed for the application of dye removals with three representative dyes (Reactive Black 5, Reactive Orange 16 and Acid Yellow 17). The perm-selective TFC-NF-R30 membrane showed a good quality of permeate with dye removal was up to 97%. Moreover, the fluxes ($1.9\text{--}11.4 \times 10^{-6} \text{ m}^3/\text{m}^2\text{s}$) were comparable to those observed with commercial membranes. These findings provide an encouraging platform for applying membrane technology in treating dye wastewater.

Keywords: Dye removal; Nanofiltration; Reaction time; Thin-film composite

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