## Desalination and Water Treatment www.deswater.com

doi: 10.1080/19443994.2015.1030710

57 (2016) 9136–9147 April



## Photocatalysis of THM precursors in reclaimed water: the application of TiO<sub>2</sub> in UV irradiation

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Received 24 October 2014; Accepted 11 March 2015

## **ABSTRACT**

In this study, ultraviolet (UV) irradiation followed by chlorination was employed for reclaimed water disinfection. In order to reduce trihalomethanes (THMs) from reclaimed water, suspended  ${\rm TiO_2}$  (10 mg/L) was added as photocatalyst in UV process to enhance the removal of THM precursors. Reduction of UV absorbance in 254 nm (UV<sub>254</sub>), dissolved organic carbon (DOC), and THMs formation was analyzed under different experimental conditions (exposure time, pH,  ${\rm TiO_2}$  doses, and  ${\rm TiO_2}$  forms). Excitation–emission matrix spectra technology was also used to investigate the changes of dissolved organic matters properties during UV and UV- ${\rm TiO_2}$  process. Expansion of irradiation time resulted in a remarkable decrease in  ${\rm UV_{254}}$  and THM yields, but showed few influence on DOC removal. THMs yield decreased more than 50% with pH increased from 5 to 9 and rise in  ${\rm TiO_2}$  dosage also presented a positive effect on photocatalytic disinfection. In addition, a dramatic increase in removal rates of  ${\rm UV_{254}}$ , DOC, and THMs was observed when  ${\rm TiO_2}$  doses were increased from 3 to 15 mg/L. In terms of  ${\rm TiO_2}$  form, suspended  ${\rm TiO_2}$  exhibited a better removal capacity on  ${\rm UV_{254}}$ , DOC, and THMs by contrast with  ${\rm TiO_2}$  coated on granular active carbon.

Keywords: UV; TiO2; THMs; Photocatalysis; Chlorination

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